



Phosphorus bioavailability from ash-rich biochars produced at different pyrolysis temperatures

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TOWARDS HUMAN AND ENVIRONMENTAL SYMBIOSIS USING BIOCHAR
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Introduction

- Peak phosphorus?
- Animal wastes and high quality biosolids as a source of P?
- The conversion of manure and biosolids into biochars? Added value?
- The effect of pyrolysis on P bioavailability? Types of feedstock and pyrolysis conditions?

Introduction

Aims:

- (i) study the **bioavailability of P** in biochars produced from biosolids (BS) and cattle manure (MA) (P rich organic wastes) at different pyrolysis temperatures;
- (ii) develop a robust **chemical method** for biochar P availability measurements.

Materials and Methods

Materials:



Two different feedstocks

Four different pyrolysis temperatures

250, 350, 450, and

550° C



Methods and Materials

Methods:

- (i) chemical analysis – including total P (by 10% HCl:HNO₃ mixture) and extractable P (2% citric acid, 2% formic acid, and neutral ammonium citrate extraction) (*standard methods for inorganic fertilizer P test*)
- (ii) a bioassay test using rye-grass grown in a P deficient sandy soil were used to compare the P bioavailability of different biochars

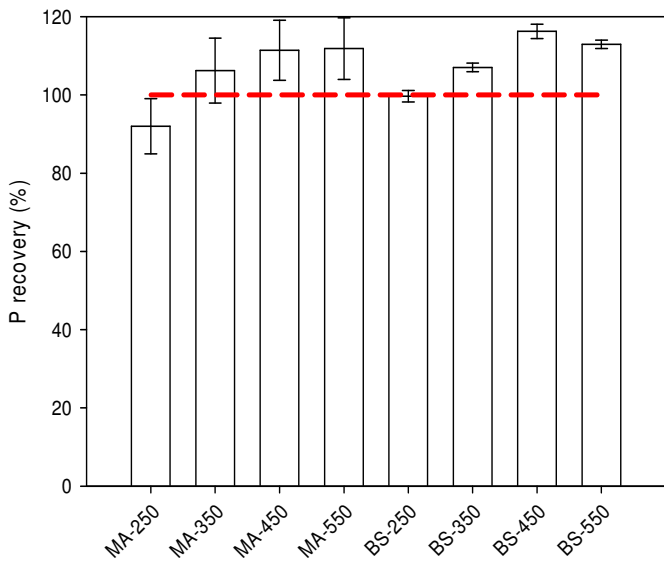
Methods and Materials

some facilities used in this studies



Results and Discussion

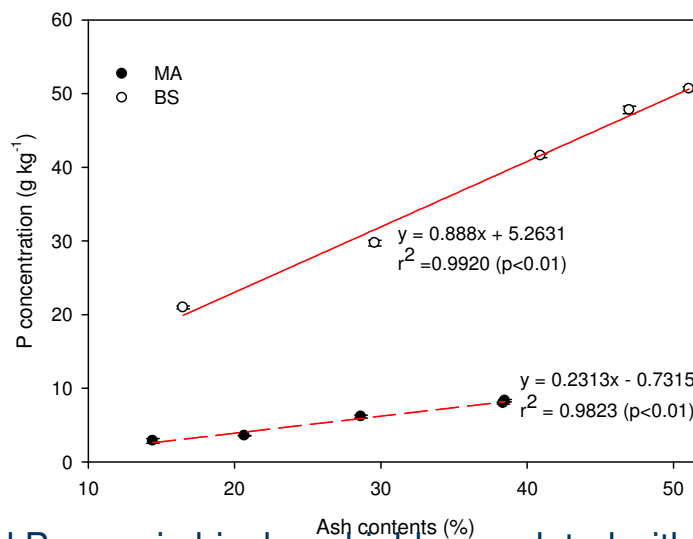
Recovery



- ~93-119% of P from feedstocks could be recovered in biochars
- High temperature biochars showed higher total P recovery than low temperature biochars due to lower recovery of organic P (more abundant in the feedstock) compared to inorganic P using this digestion procedure

Results and Discussion

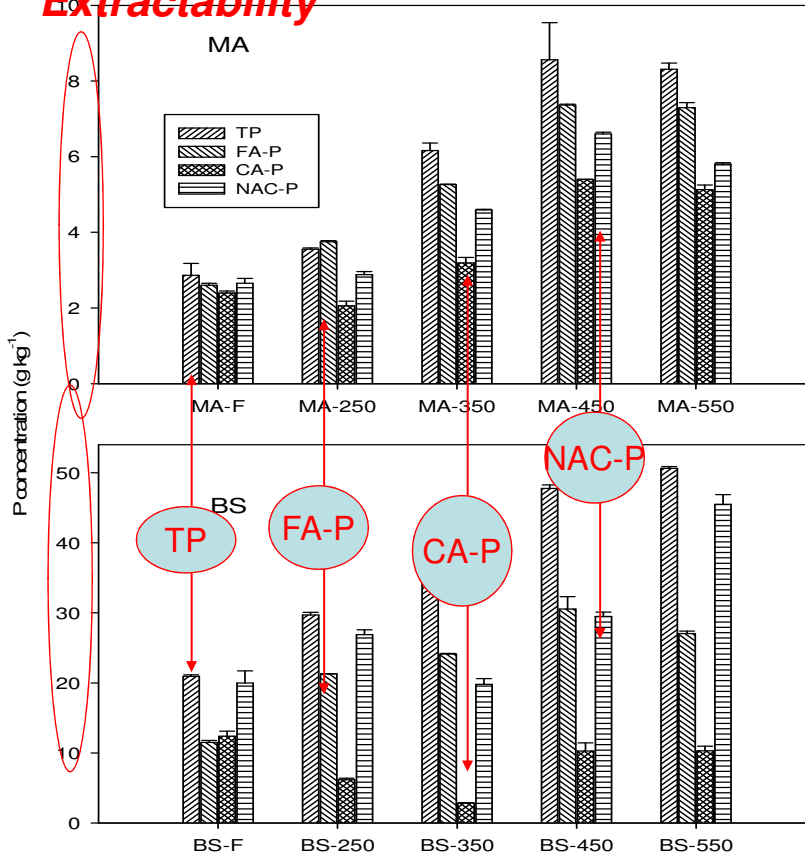
Where is it?



- Total P conc. in biochars highly correlated with ash contents
- P was enriched in **ash fractions** in biochars → the validity of the proposed method (P existed mainly in inorganic forms in biochars)

Results and Discussion

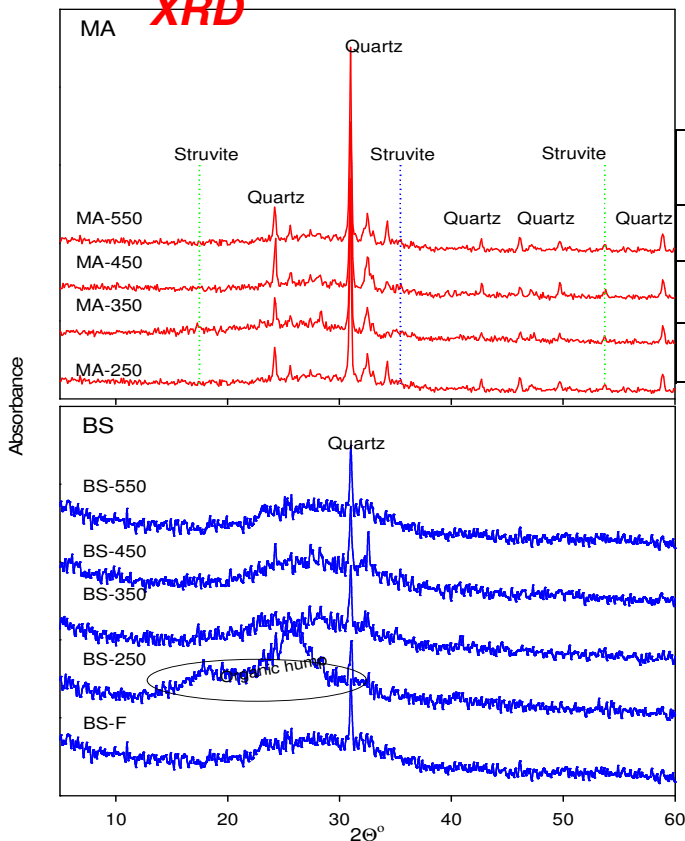
Extractability



- The fraction of extractable P in MA biochar (>69.6%) is higher than in BS biochars (<59.2%), but still more P available in biosolids
- Acidification was the main mechanism for P dissolution in MA biochars, while complexing played a key role in P dissolution in BS biochars

Results and Discussion

XRD



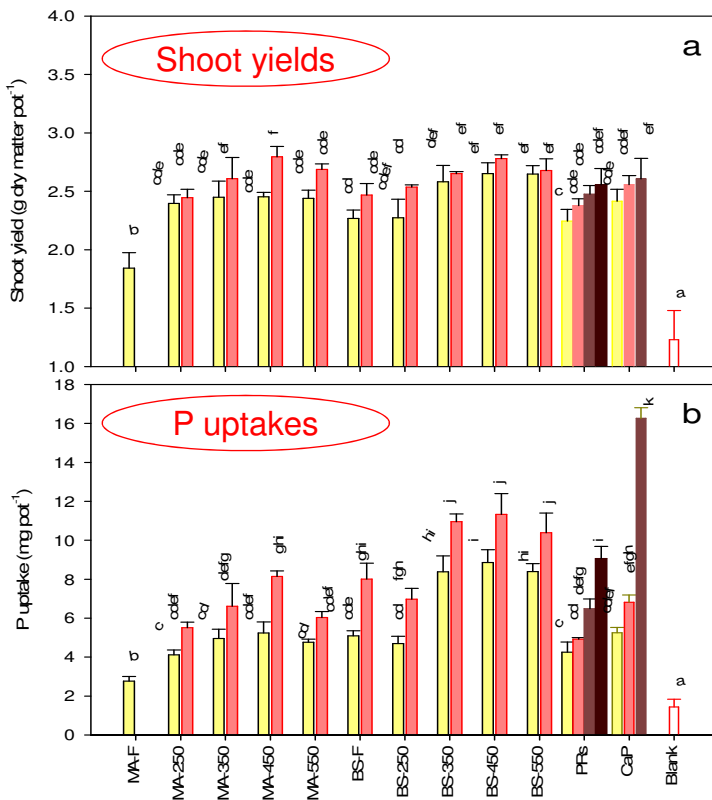
Stepwise regression

Feedstocks	Variables entered	r ²	Probability level p
MA	Mg	0.643	<0.01
BS	Al	0.865	<0.01
	Al, Mg	0.901	<0.01

Mg-P complexes dominated in MA feedstock and biochars, and amorphous Al and/or Mg-P dominated in BS feedstock and biochars.

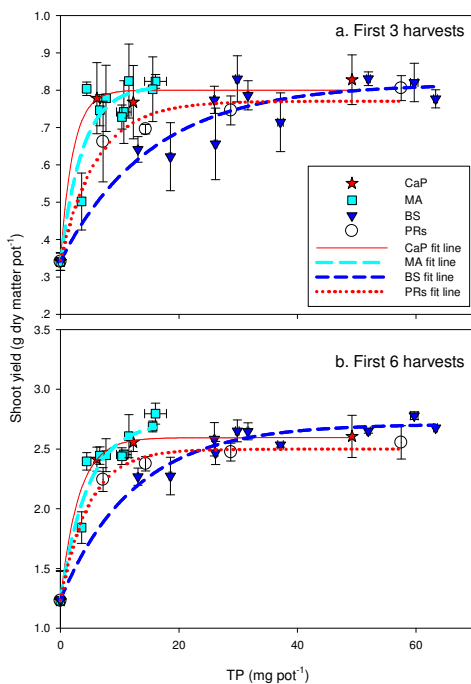
Results and Discussion

Bioassay



- Better than the control treatment and feedstocks
- P uptake was more sensitive to the dose and type of P source than plant yield
- Both dry matter yield and P uptake for both MA-550 and BS-550 decreased, compared with MA-450 and BS-450. Critical point of temperatures?

Results and Discussion



Mitscherlich equation

$$Y = Y_0 + \Delta Y - \Delta Y e^{-\epsilon N_f}$$

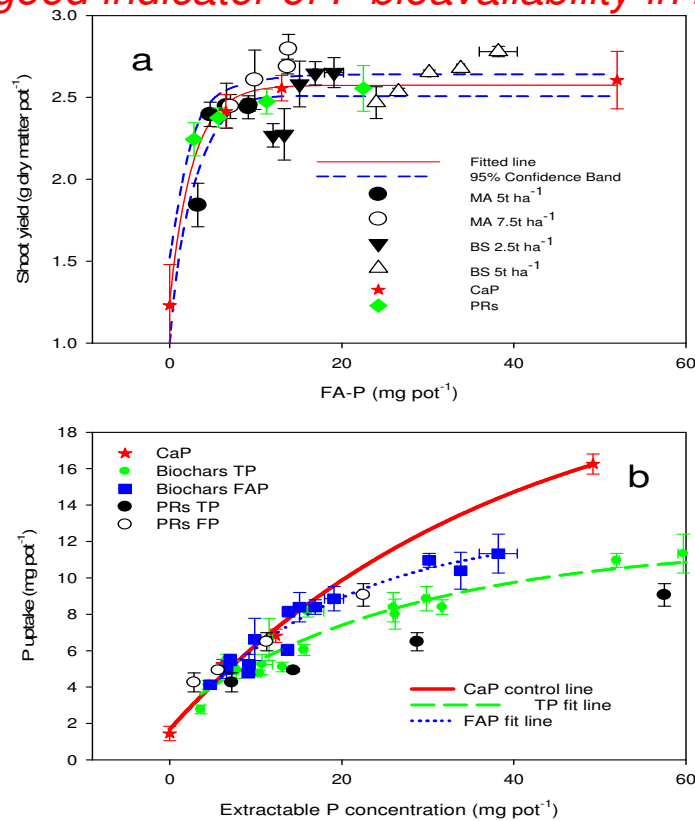
Where Y = The total dry matter yield or P uptake;
 Y_0 = The yield or P uptake without biochars or fertilizers application;
 ΔY = The difference between the maximum yield or P uptake Y_{max} and Y_0 , that is, $Y_{max} = Y_0 + \Delta Y$;
 ϵ = an activity coefficient, which is a measure for the P availability of biochars or fertilizers to the plants;
 the higher the value, the more readily available of P
 N_f = the rate of P applied

The activity coefficient (ϵ) followed a descending order as **CaP > MA > PRs > BS**. Feedstock dependent!

6 harvests yields	r^2	ϵ
CaP	0.997	0.326
MA	0.923	0.217
BS	0.966	0.081
PRs	0.993	0.209

Results and Discussion

FA-P was a good indicator of P bioavailability in biochars



Summary

- P in feedstock was **fully recovered** in the biochar and was enriched in the **ash** fraction.
- **Amorphous Al phosphate** and/or adsorbed P may dominate in **BS** biochars and **Mg phosphate** in **MA** biochars.
- After 6 harvests, the biochars were as effective as the P fertilizers tested [phosphate rocks and Ca(H₂PO₄)₂], when comparing them on a **formic acid-extractable** P basis, in increasing the shoot yield and supplying P to the plants.
- Based on the Mitscherlich equation, 2% formic acid extractable P was the most sensitive indicator of P bioavailability in biochars.

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Thank you!